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of Matter ~~PHYSICAL  
AND CHEMICAL~~

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and Chemical~~

~~Properties of~~

~~Matter~~ What's My

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vs. Physical  
changes

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Properties Physical

vs Chemical

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Matter flashcards.

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games, and other

study tools. ... This is a change in both form and nature of substance in which bonds are broken.

... In any chemical process, matter is neither lost nor gained.

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to become; symbol  
between the  
reactants and  
products in a  
chemical equation.

Law of

Conservation of  
Matter. a scientific  
law which states  
that mass/matter  
cannot be created  
or destroyed; in a

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Chemical

Nature Of

Chemical reaction,  
the total mass of  
the reactants

equals the total

mass of the

products. Balanced  
equation.

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The Nature of

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...

Atoms of elemental substances are themselves

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complex structures composed of more fundamental particles called protons, neutrons, and electrons.

Experimental evidence indicates that, within an atom, a small nucleus, which generally contains both protons and neutrons, is

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surrounded by a  
swarm, or cloud, of  
electrons.

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Chemical element -

The atomic nature  
of the elements ...

Burning and rusting  
are examples of a  
chemical property.

Law of

Conservation of

Mass. The principle  
stating that matter

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Chemical

is not created or destroyed, but can change in form.

Because matter is neither created nor destroyed, the total mass of the material(s) before the reaction is the same as the total mass of material(s) after the reaction.

Chemical Nature of

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Matter (5-PS1-2)

Chemical or

Physical Change ...

Life Science

(5-LS1-1) Plant

Growth (5 ...

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Nature Of

Matter  
All properties of matter are either extensive or

intensive and either physical or chemical.

Extensive properties, such as mass and volume, depend on the amount of matter that is being measured.

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Chemical

Nature Of

properties, such as density and color,

do not depend on the amount of

matter. Both extensive and intensive

properties are physical properties, which means they can be measured without changing the substance's

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Nature Of  
Matter  
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Chemical identity.

Matter

Physical and

Chemical

Properties of

Matter | Boundless

...

D) Amount of matter produced by the producers should be negligible when compared to the consumers. Matter

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and energy are lost with each successive step of the food chain.

Therefore, for a food web to be functional and successful, the total amount matter produced by the producers be more than that of the consumers they support.

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Nature Of

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D the chemical

properties of  
matter 3. If a

substance is

producing sound or  
new light, it is

probably

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Nature Of  
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undergoing a  
chemical reaction.  
The explosion of  
fireworks produces  
sound and light.

Based on this  
information, what  
is a likely  
conclusion ? A The  
explosion of  
fireworks is an  
example of a ch  
emical reaction . . . .

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Structure Of  
Properties of  
Matter  
Matter

Most matter  
consists of an  
agglomeration of  
molecules, which  
can be separated  
relatively easily.  
Molecules, in turn,  
are composed of  
atoms joined by  
chemical bonds  
that are more

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Chemical

Nature Of Matter  
difficult to break.

Each individual  
atom consists of  
smaller

particles—namely,  
electrons and  
nuclei.

atom | Definition,  
Structure, History,  
Examples, Diagram

...

Session 2 The  
Particle Nature of

*Page 38/42*

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Matter: Solids,  
Liquids, and Gases.

... In this session,  
participants extend  
the particle model  
by looking inside  
the particles, learn  
about some early  
chemical pioneers,  
and in the process  
discover how the  
law of conservation  
of matter applies  
even at the scale of

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Nature of  
atoms and  
molecules.

Matter

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Molecular Nature of  
Matter (6th ...

THE test described  
as new by Mr.

Wood in NATURE of

February 21 has  
been in use in my  
laboratory during  
the past year,

where it forms part  
of the regular  
routine tests  
applied to flour.

While I am ...

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